Stereoselective Total Synthesis of (–)-Kumausallene

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A stereoselective total synthesis of (–)-kumausallene was completed in 12 steps from acetylacetone. The hidden symmetry of (–)-kumausallene was recognized, and its skeleton was constructed efficiently from a C_2 -symmetric diol by a palladium-catalyzed cascade reaction. High diastereoselectivity was observed for the DMF-promoted biomimetic 1,4-bromocyclization of a conjugated enyne.

(–)-Kumausallene 1¹ belongs to a family of nonisoprenoid sesquiterpenes with a unique bromoallene moiety (Figure 1).^{2,3} The first total synthesis of (\pm)-kumausallene was accomplished by Overman.^{4–6} In this synthesis, the complex dioxabicyclo[3.3.0]octane core was constructed by an elegant ring-expansion annulation strategy^{4,5} and the bromoallene moiety was introduced by stereoselective

(4) Brown, M. J.; Harrison, T.; Herrinton, P. M.; Hopkins, M. H.; Hutchinson, K. D.; Mishra, P.; Overman, L. E. J. Am. Chem. Soc. **1991**, *113*, 5365.

(5) Brown, M. J.; Harrison, T.; Overman, L. E. J. Am. Chem. Soc. 1991, 113, 5378.

(6) Grese, T. A.; Hutchinson, K. D.; Overman, L. E. J. Org. Chem. 1993, 58, 2468.

 $S_N 2'$ displacement of propargylic sulfonate.⁷ In the first enantioselective synthesis of (–)-kumausallene by Evans,⁸ the 2,5-*cis*-substituted tetrahydrofuran (THF) ring was prepared by an acyl radical cyclization⁹ and a very efficient biomimetic strategy was undertaken to forge the bromoallene and the adjacent THF ring simultaneously.¹⁰ Although high stereoselectivity was observed for the formation of

(10) For recent reviews on biomimetic synthesis, see: (a) Scholz, U.; Winterfeldt, E. *Nat. Prod. Rep.* **2000**, *17*, 349. (b) Nicolaou, K. C.; Montagnon, T.; Snyder, S. A. *Chem. Commun.* **2003**, 551. (c) Yoder, R. A.; Johnston, J. N. *Chem. Rev.* **2005**, *105*, 4730. (d) Bulger, P. G.; Bagal, S. K.; Marquez, R. *Nat. Prod. Rep.* **2008**, *25*, 254.

(11) For synthetic studies related to kumausallene, see: (a) Osumi, K.;
Sugimura, H. Tetrahedron Lett. 1995, 36, 5789. (b) Lee, E.; Yoo, S. K.;
Cho, Y. S.; Cheon, H. S.; Chong, Y. H. Tetrahedron Lett. 1997, 38, 7757.
(c) Andrey, O.; Glanzmann, C.; Landais, Y.; Parrarapado, L. Tetrahedron 1997, 53, 2835. (d) Martin, T.; Soler, M. A.; Betancort, J. M.;
Martin, V. S. J. Org. Chem. 1997, 62, 1570. (e) Boukouvalas, J.; Fortier,
G.; Radu, Ii J. Org. Chem. 1998, 63, 916. (f) De La Pradilla, R. F.;
Montero, C.; Priego, J.; Martinez-Cruz, L. A. J. Org. Chem. 1998, 63, 9612. (g) Lee, E.; Yoo, S. K.; Choo, H.; Song, H. Y. Tetrahedron Lett.
1998, 39, 317. (h) Mereyala, H. B.; Gadikota, R. R. Tetrahedron Lett.
1998, 39, 317. (h) Mereyala, H. B.; Gadikota, R. R. Tetrahedron Lett.
1998, 39, 317. (h) Mereyala, H. B.; Gadikota, R. R. Tetrahedron, L. J. Org. Chem. 2001, 66, 1420. (j) Gadikota, R. R.; Callam, C. S.; Lowary,
T. L. J. Org. Chem. 2001, 66, 9046. (k) Chandler, C. L.; Phillips, A. J. Org. Lett. 2005, 7, 3493. (l) De La Pradilla, R. F.; Alhambra, C.;
Castellanos, A.; Fernandez, J.; Manzano, P.; Montero, C.; Urena, M.;
Viso, A. J. Org. Chem. 2005, 70, 10693. (m) Nesbitt, C. L.; Mcerlean, C. S. S. P. Tetrahedron Lett. 2009, 50, 6318.

[†] School of Pharmacy.

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⁽¹⁾ For isolation of kumausallene, see: Suzuki, T.; Koizumi, K.; Suzuki, M.; Kurosawa, E. *Chem. Lett.* **1983**, 1639.

⁽²⁾ For reviews on allene-containing natural products, see: (a) Hoffmann-Röder, A.; Krause, N. *Angew. Chem., Int. Ed.* **2004**, *43*, 1196. (b) Dembitsky, V. M.; Maoka, T. *Prog. Lipid Res.* **2007**, *46*, 328.

⁽³⁾ For recent reviews on allene synthesis, see: (a) Brummond, K. M.; Deforrest, J. E. *Synthesis* **2007**, 795. (b) Yu, S. C.; Ma, S. M. *Chem. Commun.* **2011**, *47*, 5384.

⁽⁷⁾ This strategy was also used in the synthesis of several other bromoallene-containing natural products. (a) Feldman, K. S.; Mechem, C. C.; Nader, L. J. Am. Chem. Soc. **1982**, 104, 4011. (b) Crimmins, M. T.; Emmitte, K. A. J. Am. Chem. Soc. **2001**, 123, 1533. (c) Boukouvalas, J.; Pouliot, M.; Robichaud, J.; Macneil, S.; Snieckus, V. Org. Lett. **2006**, 8, 3597. (d) Wang, J.; Pagenkopf, B. L. Org. Lett. **2007**, 9, 3703. (e) Park, J.; Kim, B.; Kim, H.; Kim, S.; Kim, D. Angew. Chem., Int. Ed. **2007**, 46, 4726. (f) Jeong, W.; Kim, M. J.; Kim, H.; Kim, S.; Kim, D.; Shin, K. J. Angew. Chem., Int. Ed. **2010**, 49, 752.

⁽⁸⁾ Evans, P. A.; Murthy, V. S.; Roseman, J. D.; Rheingold, A. L. Angew. Chem., Int. Ed. 1999, 38, 3175.

^{(9) (}a) Evans, P. A.; Roseman, J. D. *Tetrahedron Lett.* **1995**, *36*, 31.
(b) Evans, P. A.; Roseman, J. D.; Garber, L. T. J. Org. Chem. **1996**, *61*, 4880.

both THF rings, two diastereomeric bromoallenes were produced in a 1:2.5 ratio favoring the unnatural epimer. We herein describe the first diastereo- and enantioselective total synthesis of (–)-kumausallene employing a desymmetrization strategy for the construction of the 2,5-*cis*-substituted THF ring, which is distinctly different from two previous syntheses.¹¹ Only one stereoisomeric bromoallene was obtained from the biomimetic 1,4-bromoetherification of the proposed enyne precursor in the biosynthesis.



Figure 1. Representative natural products with a bromoallene moiety.

In addition to Evans' approach to (-)-kumausallene, the biomimetic strategy has also been executed in the synthesis of panacene $2^{12,13}$ and laurallene 3.¹⁴ Low diastereometic ratios (1:1 to 1.2:1) were generally observed for the newly generated stereogenic center and the axially chiral allene moiety.¹⁵ However, Canesi observed a dramatic solvent effect for 1,4-bromoetherification of a *trans*-enyne in his synthesis of (\pm) -panacene. High *syn*-diastereoselectivity was observed in chloroform (dr = 9:1), cyclohexane (dr > 20:1), and benzene (dr > 20:1), while nearly equal amounts of both diastereomets were obtained in acetonitrile or ethyl acetate.¹³ Unfortunately, the natural panacene has the *anti*-stereochemistry as shown in Figure 1. Thus, a diastereoselective biomimetic synthesis remains to be developed for this class of halogenated natural products with axial chirality.

Scheme 1. Retrosynthetic Analysis for (-)-Kumausallene



(12) Feldman, K. S. Tetrahedron Lett. 1982, 23, 3031.

(14) (a) Ishihara, J.; Shimada, Y.; Kanoh, N.; Takasugi, Y.; Fukuzawa, A.; Murai, A. *Tetrahedron* **1997**, *53*, 8371. (b) Crimmins, M. T.; Tabet, E. A. J. Am. Chem. Soc. **2000**, *122*, 5473.

We recently discovered that the diastereoselectivity of 1,4-bromolactonization of conjugated enynes depended on the choice of catalyst and *syn*-addition was generally favored.^{16,17} We envisioned that the stereochemical outcome of the bromoetherification of conjugated enynes could also be catalyst-dependent. If a 1,4-*syn*-bromoetherification of enyne **4** is realized, the synthesis of **1** can be accomplished very efficiently according to the strategy shown in Scheme 1. The upper enyne and lower alkene side chains in tetrahydrofuran **4** will be derived from functionalizations of the vinyl and lactone moieties in compound **5**. This 5-5-*cis*-fused bicyclic system can be prepared from C_2 -symmetric diol **6** by adopting a palladium-catalyzed oxycarbonylation–lactonization cascade reaction first described by Semmelhack.¹⁸

Our synthesis began with the known C_2 -symmetric diol **6**,¹⁹ available in three steps and 96% *ee* from acetylacetone.²⁰ The palladium-catalyzed cascade reaction¹⁸ provided bicyclic lactone **5** in 87% yield as a single stereoisomer (Scheme 2). Ozonolysis, Sakurai allylation,²¹ and cross-metathesis with Grubbs II catalyst²² then furnished alcohol **7** with a complete carbon framework of the lower side chain.^{11k} The Sakurai allylation proceeded with a diastereoselectivity of a 4:1 ratio favoring the desired stereoisomer.²³ The minor isomer was easily removed by column chromatography. Protection of the free hydroxyl group in compound **7** and reduction of the lactone to lactol, followed by olefination with commercially available reagent **9**, provided a separable mixture (E/Z = 10:1) of conjugated enynes.²⁴ Removal of the TMS group in basic methanol smoothly furnished the desired product **4**.

Scheme 2. Stereoselective Synthesis of Trisubstituted THF Ring by Desymetrization of a C_2 -Symmetric Diol



(16) Zhang, W.; Xu, H.-D.; Xu, H.; Tang, W. J. Am. Chem. Soc. 2009, 131, 3832.

(17) Zhang, W.; Zheng, S.; Liu, N.; Werness, J. B.; Guzei, I. A.; Tang,
 W. J. Am. Chem. Soc. 2010, 132, 3664.

(18) (a) Semmelhack, M. F.; Bodurow, C.; Baum, M. Tetrahedron Lett. **1984**, 25, 3171. (b) Semmelhack, M. F.; Kim, C.; Zhang, N.; Bodurow, C.; Sanner, M.; Dobler, W.; Meier, M. Pure Appl. Chem. **1990**, 62, 2035. For related work, see:(c) Tamaru, Y.; Kobayashi, T.; Kawamura, S.; Ochiai, H.; Hojo, M.; Yoshida, Z. Tetrahedron Lett. **1985**, 26, 3207.

(19) Whitehead, A.; Mc Reynolds, M. D.; Moore, J. D.; Hanson, P. R. Org. Lett. 2005, 7, 3375.

⁽¹³⁾ Sabot, C.; Berard, D.; Canesi, S. Org. Lett. 2008, 10, 4629.

⁽¹⁵⁾ For a model study, see: Braddock, D. C.; Bhuva, R.; Perez-Fuertes, Y.; Pouwer, R.; Roberts, C. A.; Ruggiero, A.; Stokes, E. S. E.; White, A. J. P. *Chem. Commun.* **2008**, 1419.

With envne 4 in hand, we then studied the biomimetic bromoetherification reaction. To our surprise, treatment of envne 4 with NBS in CDCl₃ led to no reaction after 24 h. Switching the solvent from CDCl₃ to toluene led to the formation of a complex mixture in 2 h. When we tried the conditions used by Evans in his biomimetic cyclization of a related substrate,^{8,25} all olefin signals completely disappeared after 6.5 h and a complex mixture was obtained. No improvement was observed when newly crystallized NBS or freshly prepared TBCD was used. We then explored different additives and solvents for the bromoetherification of envne 4. Among the additives that were able to promote the bromolactonization of conjugated enynes in our previous studies,^{16,17} only DMF and HMPA could promote the bromoetherification of conjugated enyne 4. The ratios of bromoallene products to envne starting material 4 were 4:1 and 1:2 after 24 h for DMF and HMPA respectively. However, the ¹³C NMR of the crude product showed three characteristic central allene carbons in a ratio of 1:3:1 for DMF and 2:4:1 for HMPA. In both cases, the major product among these three inseparable isomers did not match the desired bromoallene, an intermediate previously reported in Overman's synthesis.⁶ We also tried the bromoetherification for the envne substrate with a free diol derived from compound 4 by removing the TBS protecting group. However, the trans-olefin on the lower side chain participated in the halogen mediated reaction and led to unidentifiable products under various conditions.

Since deacetylkumausyne 11 (Scheme 3) was isolated together with kumausallene 1 and was proposed as the precursor for the biosynthesis of the latter,²⁶ we decided to test the biomimetic cyclization of the former. The conversion of a homoallylic alcohol to the corresponding bromide was known to be problematic and low yielding.^{5,6,8} We were able to obtain an \sim 50% yield of the homoallylic bromide 10 from alcohol 7 if all reagents (2,6-di-tertbutylpyridine, Ph_3P , and CBr_4) were carefully purified and dried (Scheme 3). Following the same sequence used previously (8 to 4), lactone 10 was converted to envne 11 in three steps. When we treated compound 11 with NBS and 10 equiv of DMF in deuterated toluene, the biomimetic 1,4-syn-bromoetherification of the conjugated envne occurred smoothly to afford final product 1 without any evidence of other byproducts. In a larger scale reaction, the amount of DMF was lowered to 1.0 equiv and product **1** was isolated in 77% yield. We found the addition of DMF to be critical because, in its absence, the reaction of enyne **11** with NBS led to a complex mixture in 1.5 h without the formation of any bromoallene.





The spectroscopic data (¹H NMR, ¹³C NMR, IR) of our synthetic (–)-kumausallene were identical in all respects to values reported in the literature.²³ The stereochemistry of the newly formed tetrahydrofuran ring was further confirmed by the nuclear Overhauser effect (NOE).²³ The observed negative sign of optical rotation ($[\alpha]_D^{20} = -141$ (c = 0.24, CHCl₃); lit¹ $[\alpha]_D^{20} = -150$ (c = 1, CHCl₃); lit⁸ $[\alpha]_D^{20} = -145$ (c = 0.4, CHCl₃)) provided further evidence for the absolute stereochemistry of the bromoallene^{2,17} based on Lowe's rule.²⁷

In summary, we have completed the first diastereo- and enantioselective synthesis of (-)-kumausallene in 12 steps from acetylacetone. We recognized the hidden symmetry of the target and prepared the 2,5-*cis*-substituted THF ring by desymmetrizing a C_2 -symmetric diol using a palladium catalyzed cascade reaction. A stereoselective biomimetic cyclization was realized when the 1,4-bromoetherification was conducted on the proposed enyne precursor in biosynthesis.

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Supporting Information Available. ¹H NMR, ¹³C NMR, IR, HRMS, and optical rotations for intermediates involved in the synthesis. This material is available free of charge via the Internet at http://pubs.acs.org.

^{(20) (}a) Rychnovsky, S. D.; Griesgraber, G.; Zeller, S.; Skalitzky, D. J. J. Org. Chem. **1991**, 56, 5161. (b) Rychnovsky, S. D.; Griesgraber, G.; Powers, J. Org. Synth., Coll. Vol. 10 **2004**, 276.

^{(21) (}a) Hosomi, A.; Endo, M.; Sakurai, H. *Chem. Lett.* **1976**, 941. (b) Hosomi, A.; Sakurai, H. *Tetrahedron Lett.* **1976**, 1295.

⁽²²⁾ Scholl, M.; Ding, S.; Lee, C. W.; Grubbs, R. H. Org. Lett. 1999, 1, 953.

⁽²³⁾ See Supporting Information for details.

⁽²⁴⁾ Corey, E. J.; Ruden, R. A. Tetrahedron Lett. 1973, 1495.

⁽²⁵⁾ Condition used by Evans: tetrabromocyclohexadienone (TBCD) in methylene chloride.

⁽²⁶⁾ Suzuki, T.; Koizumi, K.; Suzuki, M.; Kurosawa, E. Chem. Lett. **1983**, 1643.

⁽²⁷⁾ Lowe, G. J. Chem. Soc., Chem. Comm. 1965, 411.